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The equilibrium constant - chemwiki: the dynamic

The equilibrium constant, K Notice the mathematical product of the chemical products raised to the which is the inverse of the autoionization constant of

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The concept of chemical equilibrium was developed after Berthollet For example, in the case of acetic acid dissolved in water and forming acetate and hydronium ions,

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the equilibrium of acid dissociation can be written symbolically as: Acid dissociation constants are also essential in aquatic chemistry and chemical

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Chemical equilibrium has been reached in a reaction when the rate of the forward reaction is Pure liquids do not appear in the equilibrium expression. Water

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Although K_w is defined in terms of the dissociation of water, this equilibrium constant expression is equally valid for solutions of acids and bases dissolved in water.

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